

7.17 (1H, *d*, *J* = 8 Hz, H-6'), 8.20 (1H, *s*, H-2), 8.30 (1H, *d*, *J* = 8 Hz, H-5);  $^{13}\text{C}$  NMR [25.2 MHz,  $(\text{CD}_3)_2\text{SO}$ ]\*: 60.5 (OMe), 102.6 (C-8), 107.7 (C-5'), 111.8 (C-1'), 115.7 (C-6), 116.8 (C-4a), 122.2 (C-3), 126.1 (C-6'), 127.6 (C-5), 136.3 (C-3'), 149.5† (C-2'), 151.1† (C-4'), 154.8 (C-2), 157.9 (C-8a), 162.9 (C-7), 176.0 (C=O); EIMS (probe) 70 eV, *m/z* (rel. int.): 300 [*M*]<sup>+</sup> (100), 285 (30), 282 (50), 254 (20), 164 (10), 137 (30).

7,2',4'-Triacetoxy-3'-methoxyisoflavone. Mp 151–153° from petrol– $\text{CHCl}_3$ ;  $^1\text{H}$  NMR [60 MHz,  $(\text{CD}_3)_2\text{CO}$ ]:  $\delta$  2.10 (3H, *s*, 1 OAc), 2.30 (6H, *s*, 2 OAc), 3.80 (3H, *s*, OMe), 7.10–7.50 (4H, *m*, H-6, H-8, H-5', H-6'), 8.17 (1H, *s*, H-2), 8.2 (1H, *d*, *J* = 8 Hz, H-5'); EIMS (probe) 70 eV, *m/z* (rel. int.): 426 [*M*]<sup>+</sup> (5), 384 (90), 342 (100), 300 (30), 282 (80), 164 (5), 137 (15).

\* $\delta$  (TMS) =  $\delta$   $(\text{CD}_3)_2\text{SO}$  + 39.5 ppm.

†Signals may be interchanged.

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STRICTAMINE-N-OXIDE FROM *RHAZYA STRICTA*

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**Key Word Index**—*Rhazya stricta*; Apocynaceae; alkaloids; strictamine; NMR.

**Abstract**—A new alkaloid, strictamine-*N*-oxide has been isolated from the leaves of *Rhazya stricta*.

## INTRODUCTION

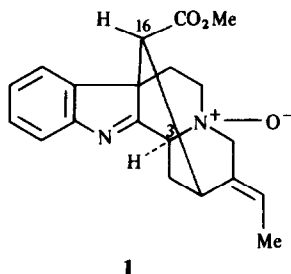
*Rhazya stricta* is a small glabrous erect shrub which is abundantly distributed in various parts of Pakistan [1–3] and is reported in the indigenous medicinal system as a curative for chronic rheumatism and as a bitter tonic for sore throats and in fever [4, 5]. Extracts of *R. stricta* showed a marked leucopenic effect in rats when given orally and a number of cytotoxic alkaloids have been reported from this plant [6, 7].

## RESULTS AND DISCUSSION

Chromatography of the alcoholic extracts of the leaves afforded a substance which gave a UV spectrum typical of the indolenine chromophore. The IR spectrum ( $\text{CHCl}_3$ ) showed the presence of an ester carbonyl group. High resolution mass spectrometry afforded the [*M*]<sup>+</sup> at *m/z* 338.1625 which corresponded closely with the mass calculated for the formula  $\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3$  (338.1630) while the rest of the spectrum was very similar to that reported for strictamine [8, 9].

The  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ ) showed a three-proton double doublet at  $\delta$  1.625 ( $J_1 = 7$  Hz,  $J_2 = 2.5$  Hz) assigned to the methyl of an ethylidene group. A one-proton quartet at  $\delta$  5.75 ( $J = 7$  Hz) was assigned to the olefinic proton of the ethylidene group. The ester methyl group resonated as a three-proton singlet at  $\delta$  3.73. A characteristic one-proton doublet for the C-3 proton appeared at the rather downfield position of  $\delta$  5.61 ( $J = 6.25$  Hz) on account of the deshielding influence of the vicinal quaternary nitrogen. Another doublet at  $\delta$  2.15 ( $J = 3.25$  Hz) was assigned to the C-16 proton, the upfield shift being on account of the shielding influence of the indolenine nucleus on which it overlies [6, 8].

The aromatic protons resonated as complex multiplets in the region between  $\delta$  7.21–8.77. Because of the strong similarities of the  $^1\text{H}$  NMR and the mass spectrum with those of strictamine, as well as the polar nature of the compound it was suspected that the substance isolated was strictamine-*N*-oxide. For confirmation of this the material was subjected to deoxygenation in dichloromethane with phosphorus trichloride. After 25 min at 30° complete conversion to a faster moving material on TLC



was observed. Usual work up afforded a white crystalline material which was identified as strictamine by comparison of its physical and spectral data (co-chromatography, mp,  $[\alpha]_D$ , IR, UV, NMR, mass spectrum) with those reported in the literature.

In order to ascertain that strictamine-*N*-oxide is a genuine natural product and not an artifact of isolation, strictamine was separately exposed to identical extraction and separation procedures but no conversion to the corresponding *N*-oxide was discernible. Further confirmation of this was obtained by detection of strictamine-*N*-oxide by TLC in freshly prepared crude extracts.

#### EXPERIMENTAL

**Isolation of strictamine-*N*-oxide.** The EtOH extract of the fresh leaves of *R. stricta* Decaisne was concd and taken up in 10% HOAc. The non-alkaloidal portion was removed by extraction into EtOAc. The aq. acidic soln was basified with aq.  $\text{NH}_3$  and extracted with EtOAc to afford the crude alkaloids (170 g). This alkaloidal material was subjected to preliminary fractionation by flash chromatography over a silica gel column. Elution with petrol,  $\text{C}_6\text{H}_6$ ,  $\text{CHCl}_3$ , EtOAc and finally with MeOH afforded a number of fractions. The fraction obtained on elution with petrol- $\text{CHCl}_3$  (4:5) was concd and again subjected to flash chromatography over a silica gel column, elution being carried out with  $\text{CHCl}_3$ -MeOH (43:7). This afforded a mixture of four alkaloids which were separated by CC on neutral alumina. A slower moving alkaloid thus obtained in a crude state was further

purified on precoated prep. TLC plates (Merck aluminum oxide Type EF-254) in  $\text{CHCl}_3$ -MeOH (19:1). A pure substance was thus obtained as a colourless amorphous material (12 mg),  $[\alpha]_D -75^\circ$ , which gave an orange colouration with Dragendorff's reagent and a dark pink colour with  $\text{CeSO}_4$  soln.  $[\alpha]_D$  ( $\text{CHCl}_3$ )  $-75^\circ$ ; IR  $\nu_{\text{CHCl}_3}^{\text{max}}$  1740  $\text{cm}^{-1}$ ; UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 213 (3.15), 262 (2.66);  $\lambda_{\text{max}}^{\text{MeOH}}$  238 nm (log  $\epsilon$  2.59);  $^1\text{H}$  NMR ( $\text{CDCl}_3$ );  $\delta$  1.625 (3H, dd,  $J_1 = 7$  Hz,  $J_2 = 2.5$  Hz, C=CH-Me),  $\delta$  5.75 (1H, q,  $J = 7$  Hz, C=CH-Me),  $\delta$  3.73 (3H, s, OMe),  $\delta$  5.61 (1H, d,  $J = 6.25$  Hz, C-3H),  $\delta$  2.15 (1H, d,  $J = 3.2$  Hz, C-16H),  $\delta$  7.21-8.77 (4H, m, ArH); HRMS:  $[\text{M}]^+$  338.1625 (calc. for  $\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3$ , 338.1630);  $m/z = 322.1657$  (100%, calc. for  $\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_2$ , 322.168  $[\text{M} - \text{O}]^+$ , 294, 263, 230 and 122.0971 (calc. for  $\text{C}_8\text{H}_{12}\text{N}$ , 122.0969).

**Conversion of strictamine-*N*-oxide to strictamine.** Strictamine-*N*-oxide (7 mg) was dissolved in 0.5 ml  $\text{CH}_2\text{Cl}_2$  and 0.1 ml  $\text{PCl}_3$  added. The mixture was stirred for 25 min at  $30^\circ$  and after basification with aq.  $\text{NH}_3$  to pH 14, extraction with  $\text{CHCl}_3$  afforded a white crystalline material which was identified as strictamine by comparison of its physical and spectral data (co-chromatography, mp,  $[\alpha]_D$ , IR, UV, NMR, MS) with those reported in the lit.

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